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Section 25.4 Fission
and Fusion of Atomic
Nuclei In your
textbook, read about
the process of by which
electrical energy is
produced in a nuclear

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power plant. Use the following diagram to complete the passage.

D In a nuclear power plant, energy is produced in reactor by fission reactions that occur in uranium-containing bars called (I)

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nuclear chemistry from
discovery to

application. You will
identify types of

radioactive decay and
solve decay rate

problems. You will

describe the reactions
involved in nuclear

fission and fusion. You

will learn about appli-
cations of nuclear

reactions

Chapter 25: Nuclear Chemistry

Chapter 25 Section

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25.2 (continued) Half-Life Discuss Explain that, for each element, there exists only a small range of neutron-to-proton ratios that produce stable nuclei. If a nucleus does not reflect a stable ratio, it spontaneously decays until a stable ratio of neutrons to protons results. Relate Explain that the nuclear stability that

25.2 Nuclear
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Transformations 25

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chemistry video below
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watching the video,
then read through
sections 1 and 2 to
complete the first
assignment of the
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25A .

13. Chapter 25:
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Answer Key

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Ch 4 & 25 The Atom
and Nuclear Chemistry

How Atoms Differ #1

Fill in this chart for
these Elements &

Isotopes: Element

Atomic Protons

Neutrons Electrons

Element & Symbol

13 14Si 36Cl 17 48

24Cr 38 Ar 18 17 80 39

K 19 30Zn mass 26.98

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Neutron Ejection. when
a neutron is emitted
from the nucleus. ${}^1_0\text{n}$.
Particle for Neutron
Ejection. ${}^5_2\text{He} \rightarrow {}^1_0\text{n} +$
 ${}^4_2\text{He}$.

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692 Chapter 16

Nuclear Chemistry 16.1

The Nucleus and

Radioactivity Our

journey into the center

of the atom begins

with a brief review. You

learned in Chapter 3

that the protons and

neutrons in each atom

are found in a tiny,

central nucleus that

measures about

$1/100,000$ the

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diameter of the atom
itself. You also learned

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Fission and Fusion 25.4

Radiation in Your Life.

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Transformations 25.3

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Radiation in Your Life.

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